

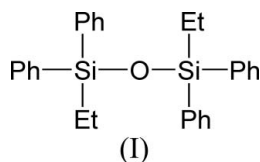
1,3-Diethyl-1,1,3,3-tetraphenyldisiloxane

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Key indicators

Single-crystal X-ray study
 $T = 200$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.052
 wR factor = 0.157
Data-to-parameter ratio = 10.0For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Molecules of the title compound, $\text{C}_{28}\text{H}_{30}\text{OSi}_2$, are centrosym-
metric, with the O atom lying on an inversion centre. The ethyl
substituents are placed *trans* along the strictly linear Si—O—
Si fragment.

Comment

Crystal structures of *ca* 40 acyclic disiloxanes are known, but a
sound explanation why their Si—O—Si angles vary within the
140–180° range still cannot be provided. The Cambridge
Structural Database (CSD, Version 5.27 of 2006 with updates;
Allen, 2002) stores data for eight symmetric disiloxanes of
formula $\text{Ph}_2\text{RSiOSiRPh}_2$; these include $R = \text{Me}$ (CSD refcode
GIMZUX; Wojnowski *et al.*, 1988), tBu (CSD refcode
DEPGOU; Karle *et al.*, 1986) and Ph [CSD refcodes
OXTPSI02 (Kooijman *et al.*, 1999) and OXTPSI10 (Glidewell
& Liles, 1978)], for which the Si—O—Si angles are 140, 152.5
and 180°, respectively. We have prepared and structurally
characterized 1,3-diethyl-1,1,3,3-tetraphenyldisiloxane
($\text{Ph}_2\text{EtSiOSiEtPh}_2$), (I), a compound which fits well in this
series. The structure of (I) is shown in Fig. 1 and selected
geometric parameters are given in Table 1.The two halves of the molecule of (I) are related by an
inversion centre located at atom O1. It follows that, despite
similarities between alkyl substituents (Me, Et and tBu), the
geometry of (I) resembles most closely that of hexaphenyl-
disiloxane; in both compounds ($R = \text{Et}, \text{Ph}$), the central Si—
O—Si fragment is strictly linear. The same holds for Si—O
bond lengths which are virtually the same in (I) and $\text{Ph}_3\text{SiO}-$
 SiPh_3 [1.616 (1) Å according to Glidewell & Liles (1978), and
1.620 (1) Å determined at 150 K by Kooijman *et al.* (1999)],
but remarkably longer in the case of $\text{Ph}_2\text{MeSiOSiMePh}_2$
[1.637 (2) and 1.640 (2) Å; Wojnowski *et al.*, 1988] and
 $\text{Ph}_2\text{tBuSiOSi}^t\text{BuPh}_2$ [1.633 (1) and 1.638 (1) Å; Karle *et al.*,
1986]. These two sets of observed values correlate with the
type of disiloxane structure, linear or bent, and the mutual
arrangement of the alkyl substituents. Unlike $\text{Ph}_2\text{EtSiO}-$
 SiEtPh_2 , (I), where the ethyl groups are *trans*, in $\text{Ph}_2\text{MeSiO}-$
 SiMePh_2 and $\text{Ph}_2\text{tBuSiOSi}^t\text{BuPh}_2$ the alkyl substituents are
gauche along the Si···Si axis (see Newman projections in
Fig. 2). This conformation is accompanied by a bending of the
Si—O—Si fragment and elongation of the Si—O bonds. TheReceived 9 January 2007
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same phenomenon is observed in other known disiloxanes of type $(\text{Ph}_2\text{RSi})_2\text{O}$ [CSD refcodes DAPNAK and DAPNOY (Nitsche *et al.*, 2003), TANKOJ (Lee *et al.*, 2002), ECOHUA and EDAFAR (Dirnens *et al.*, 2003)]; only those with *R* substituents in *trans* positions have linear Si—O—Si skeletons. The case of 1,3-bis[3-(3-pyridyl)isoxazolin-5-yl]-1,1,3,3-tetraphenyldisiloxane (Dirnens *et al.*, 2003) is especially interesting as all three types of conformers are known for this compound. The *trans* form has a linear skeleton, but the *cis* and *gauche* forms are bent.

Experimental

Crystals of (I) suitable for X-ray measurements were obtained by hydrolysis of a toluene solution of 1,3-diethyl-1,1,3,3-tetraphenyldisilthiane (Becker & Wojnowski, 1982) with an excess of aqueous AgNO_3 , followed by the usual workup and crystallization of the crude product from hexane–benzene (4:1 *v/v*).

Crystal data

$\text{C}_{28}\text{H}_{30}\text{OSi}_2$	$Z = 2$
$M_r = 438.7$	$D_x = 1.199 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 7.8479 (14) \text{ \AA}$	$\mu = 0.16 \text{ mm}^{-1}$
$b = 20.603 (3) \text{ \AA}$	$T = 200 (2) \text{ K}$
$c = 7.5978 (15) \text{ \AA}$	Block, colourless
$\beta = 98.541 (19)^\circ$	$0.22 \times 0.20 \times 0.16 \text{ mm}$
$V = 1214.9 (4) \text{ \AA}^3$	

Data collection

Kuma KM-4-CCD diffractometer	2028 independent reflections
ω scans	1590 reflections with $I > 2\sigma(I)$
Absorption correction: none	$R_{\text{int}} = 0.055$
4916 measured reflections	$\theta_{\text{max}} = 25^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0917P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.052$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.157$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.14$	$\Delta\rho_{\text{max}} = 0.35 \text{ e \AA}^{-3}$
2028 reflections	$\Delta\rho_{\text{min}} = -0.39 \text{ e \AA}^{-3}$
203 parameters	Extinction correction: <i>SHELXL97</i>
All H-atom parameters refined	Extinction coefficient: 0.023 (6)

Table 1

Selected geometric parameters (\AA , $^\circ$).

Si1—O1	1.6181 (7)	Si1—C1	1.867 (3)
Si1—C13	1.857 (3)	Si1—C7	1.870 (3)
O1—Si1—C13	109.29 (11)	Si1—O1—Si1 ⁱ	180
O1—Si1—C1	108.74 (9)	C2—C1—Si1	120.4 (2)
C13—Si1—C1	112.27 (14)	C6—C1—Si1	122.4 (2)
O1—Si1—C7	108.48 (9)	C8—C7—Si1	123.3 (2)
C13—Si1—C7	108.01 (13)	C12—C7—Si1	120.1 (2)
C1—Si1—C7	109.98 (12)	C14—C13—Si1	115.6 (2)

Symmetry code: (i) $-x + 2, -y + 1, -z + 1$.

All H atoms were found in an electron difference Fourier map and refined without constraints or restraints.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2006); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2006); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97*

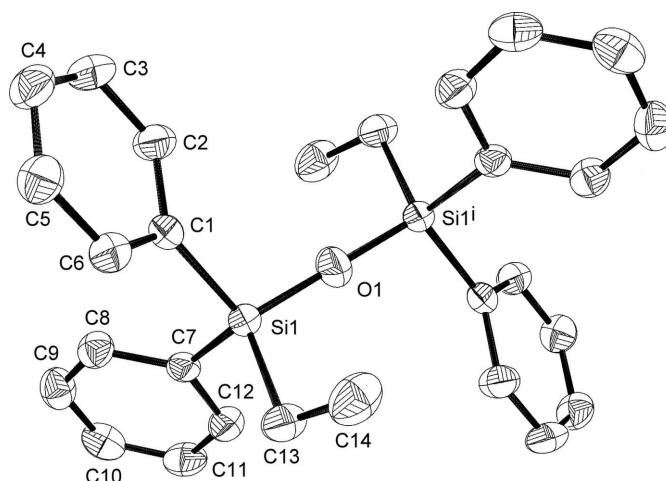


Figure 1

The molecular structure and atom-numbering scheme for (I), with displacement ellipsoids drawn at the 50% probability level. H atoms have been omitted for clarity. [Symmetry code: (i) $-x + 2, -y + 1, -z + 1$.]

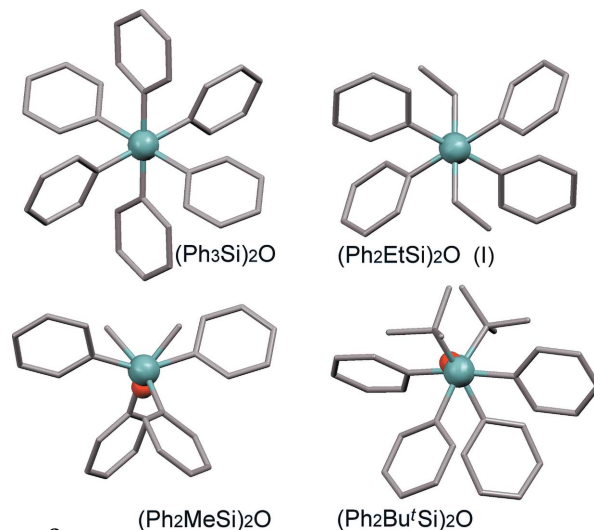


Figure 2

Newman projections along the Si···Si axis for $(\text{Ph}_2\text{RSi})_2\text{O}$ molecules ($R = \text{Me, Et, Bu}$ and Ph).

(Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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